

# Biodiesel Production from Various Feed stocks by Transesterification: A Reviewed studies

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## Abstract

Seeing that fossil fuel emissions will hold indefinitely, we must discover a appropriate and lengthy- term opportunity, thanks to the reality that it's miles biodegradable, non-toxic, and eco-friendly, biodiesel an excellent substitute for diesel engines. EASAC classifies the evolution of biodiesel into four generations. Biodiesel feedstocks and their blessings and disadvantages for distinct generations of the gasoline are thoroughly analysed in this text. An in-depth investigation is furnished in this newsletter, of the benefits and disadvantages of diverse feedstocks used inside the manufacturing technique of various generations of biodiesel. In phrases of the manufacturing of biodiesel, transesterification is the great method as it produces excessive-yield biodiesel with comparable houses to diesel, making it an ideal preference. As a ways as economics are involved, this technique is also feasible. It's far viable to satisfy the strength necessities of the future by means of mixing different oil feedstocks. The machine used and the value of feedstock have the maximum big effect on the value of biodiesel production. Traits of biodiesel consisting of the oxidation stability, bloodless drift and cetane range, viscosity, and density, are a number of the maximum critical characteristics of biodiesel. Biodiesel's performance in diesel engines changed into additionally mentioned on this paper, and it changed into counseled that biodiesel is safer for the environment than Petro-diesel. In contrast to Petro-diesel, it degrades four times faster and has with a better flash point, making storage and coping with less complicated. It is also riskless and reasons much less inflammation to the skin than soap and water. The paper also looked at the manufacturing of biodiesel using feedstocks from the first via the fourth technology.

**Key Words:** Biodiesel, transesterification

## Introduction

Biomass was the primary energy source prior to the nineteenth century, but as technology advances, the search for new energy-efficient sources of energy continues. For cooking and heating, dried wood was the most common fuel, ethanol and vegetable oil on the other hand were the most commonly used fuels for lighting. Early humans also used wood to heat their homes and to heat-treat clay. Industrialization's need for energy led to a shift over many years, the transition from wood to coal, and then from coal to fossil fuels, has occurred. As a result, fossil fuels have come to dominate the world's energy supply, accounting for about 48% of total demand.

As a result of growing concerns about environmental safety, as well as an ever-increasing global demand of fuel for usage in transportation and the industrial sector have led to an ever-demand for energy is on the rise. Because in the event of the depletion of fossil fuel reserves, has sparked a search for new renewable energy sources and Liquid fuel sources that are friendly to the environment. All forms of economic and social growth are dependent on the availability of energy, in the field of a variety sources of alternative energy, the natural phenomena process

of converting feedstock into useful forms of energy takes place. The socioeconomic benefits of using renewable energy sources instead of conventional power generation methods are significant. It is imperative that biofuels be developed for four primary reasons: (1) combating climate change (2) responding to rising energy consumption (3) ensuring a reliable supply of energy (4) and maximizing the use of limited resources. Combating global warming and exploring low- carbon fuel alternatives are two of the most important tasks we face today. The world's population is expected to grow to 8 billion or even 10 billion people by the year 2050, according to estimates. With the rise of emerging economies, the demand for energy is expected to rise significantly. As fuel resources are dispersed across the globe, meeting future energy demand will be a major challenge. Another critical aspect of our long-term survival is cutting down on the amount of energy we waste and making the most of our precious natural resources, all of these issues can be solved by biofuels.

When compared to commercial diesel fuel, the Biodiesel is considered to be one of the preferable possible sources of biofuel because it has no effect on engine performance parameters. Another benefit of using biodiesel is that it is non-toxic and oxygenated to reduce engine emissions, as well as being biodegradable and free of Sulphur and minerals.

The biochemical process by which biodiesel is produced is known as transesterification by using biological components such as animal or vegetable fats, for example. However, a century ago Rudolf Diesel experimented with the use of engine fuel derived from vegetable oil, several problems arise in the case of vegetable oils directly used in diesel engines, many issues, such as a result of incomplete combustion and excessive smoke emission as well as deposition of carbon on the engine's components as a result of its high viscosity; of SVO's, straight vegetable oils. In addition, vegetable oil has a high flash point, which reduces the oil's volatility. Using transesterification, biodiesel has been formulated to overcome these issues. Biodiesel esters and glycerol are formed when Alcohols react with fats or oils to produce biodiesel. In order to accelerate the reaction time and yield increase, an alkaline or acidic catalyst is used. Acid alkyl esters from fatty acids are formed during the transesterification of triglycerides is the goal of the process of transesterification Transformation of triglycerides into fatty acid alkyl esters is the goal of the transesterification process, by reacting an alkali or an acid with an alcohol, such as methanol, and resulting in glycerol as a waste product.

Renewable fuel research has focused on biodiesel because of its ability to be produced from biomass, Non-toxic and non-explosive properties, non-flammable, higher flash point, Application in diesel engines without the need for large changes is possible and Reduced reliance on fossil fuels. Biodiesel can be used as a "clean energy source" in place of fossil fuels. It can reduce CO<sub>2</sub>, SO<sub>2</sub>, CO, and HC emissions. In contrast to the biodiesel combustion process, plants absorb more CO<sub>2</sub> than are released. Thus, compared to the use of fossil fuel, using biodiesel can more effectively reduce CO<sub>2</sub> emissions, protect the environment, and maintain ecological balance.

Despite the fact that using edible oils to make biodiesel helps reduce our dependence on fossil fuels, there has been widespread criticism of this strategy due to the conflict between food and fuel. Despite the fact that the world's population continues to expand, there will be more people who will need to be fed. Using food crops to make biodiesel could lead to food shortages and price spikes if food demand outpaces supply.

## **Feedstock**

Vegetable oil, for example, can be utilised as a feedstock in the production of biodiesel, algae, microbial oil and animal fats. There are significant differences in the purity and compositions of biodiesel obtained from different feedstocks, A biodiesel plant's feedstock is selected as the first step in the manufacturing process, which has a significant impact on a number of variables, Such as biodiesel purity, price, composition, and production. Classifying edible and non-edible as well as waste-based biodiesel sources is dependent on the feedstocks' availability and type. Some additional considerations must also be made when selecting sources of biodiesel feedstock that are locally accessible and competitively priced and easily accessible as well as technically feasible.

## **Types of feedstocks**

### ***First generation biodiesel***

Biodiesel from the first generation is made from food-based feedstocks, edible feedstocks include things like rapeseed and soybean oils as well as coconut and palm oils as well as mustard and olive oils, etc.

At the dawn of the biodiesel era, the production of biodiesel from edible feedstocks was widespread. Plants are readily available, and the conversion process is relatively simple, making first-generation feedstocks an attractive option. Because these feedstocks raise the cost of food products, there is a risk of food shortages. Adaptability to environmental conditions, for biodiesel production from edible feedstock, high costs and having a small area to cultivate is also an issue. These drawbacks compelled biodiesel producers to look for additional sources of fuel.

### ***Second generation biodiesel***

biodiesel of Second generation is made from (nonedible) feedstocks, Nagchampa oil, Jatropha oil, Neem oil, Calophyllum oil, Karanja oil, Rubber seed oil, and Mahua Indica oil examples of non-edible feedstocks. Non-edible feedstocks are being researched because of first generation feedstocks drawbacks of. In addition to being environmentally friendly and cost-effective, second generation biodiesel also eliminates food inequity and reduces the need for farmland.

There is no need to replant food crops and no need for agricultural land when using second generation biodiesel oils. Second-generation fuels have some drawbacks, such as lower yields from (nonedible) plants like Jojoba oil, Jatropha oil, and Karanja oil, which are the main non-edible sources of these fuels.

### ***Third generation biodiesel***

It is referred to as "third generation biodiesel" when it is produced using microalgae and discarded cooking oils. decreased emissions of greenhouse gases are one of the most significant advantages of third-generation biodiesel, increased productivity and growth, less competition for farming land, more oil, and less impact on the food supply. The main drawbacks are the high costs, the need for sunlight, the difficulty of mass production, and the difficulty of extracting oil. Currently, research is being carried out in order to increase the production rate of biodiesel from algal biomass as well as the extraction process. Fish oil and animal fat are the primary sources of third-generation biodiesel, micro algae, waste cooking oil etc. This third generation of biodiesel's feasible sources overcome the problems of earlier generations' feedstocks, which affect the food chain availability and the environment. conditions of the environment, economic viability, As needed, it can be adapted to suit the requirements.

### ***Fourth generation biodiesel***

solar-powered fuels derived from photobiological processes are being considered for the biodiesel's fourth generation. By converting solar energy into biodiesel from raw materials, solar biofuels can be created, an entirely new area of study is being conducted on this conversion procedure. A wide variety of raw materials can be found. unlimited supply at a low cost. Such a transformation is made possible by the use of synthetic biology. For long-term growth and development, need to look for solutions that are brand new to nature that make synthetic living firms and modern microorganisms to directly convert solar energy and effectively converted into a source of power. Similarly, metabolically engineered microbial fuel development with inorganic water splitting catalysts is an expanding method for liquid fuel production and storage.

### **Parameters to be checked for feed stock**

#### ***Moisture***

Soaps and emulsions can form during the process of transesterification when moisture interacts with the catalyst. Less than 0.050 wt. % should be expected.

#### ***Free Fatty Acid (FFA)***

The interaction of FFA in Sodium methoxide catalyst and feedstock emulsions may be formed making the process of biodiesel separation more difficult; resulting, perhaps, in a reduction in yield. Additionally, emulsions can increase costs by requiring additional cleaning and filter replacement. To decrease the amount of soap which is generated as a result of a chemical reaction, FFA in the feedstock should be reduced to 0.5 weight percent or less as a goal.

## ***Kinematic Viscosity***

Velocity measures a substance's resistance shear or allow to flow; the Movement of liquid near a solid border, like the piping's interior walls, which described by the formula, which, to begin with, is strongly influenced by temperature.

## ***FAC Color***

Members of (FAC) a Fat Analysis Committee, the color technique compares the oils and fats to color standards to determine their color.

## ***Value of Saponification***

a measure of the amount of saponification, is defined as (KOH) potassium hydroxide, in milligrams, it is necessary in order to saponify a gram of fat or oil in the specified conditions. In terms of molecular weight, and length of the fatty acid chain in a given fatty acid that existing is used to calculate the value of saponification. The average chain length of fatty acids can be compared.

## ***Volatile Matter and Moisture***

Volatile matter in a feedstock that found, may lead to loss in fatty acid methyl ester yield either by diluting the feedstock, or by reacting with the catalyst.

## ***Insoluble Impurities***

Measurement of solids is done using an insoluble impurities tests, that are kerosene and petroleum, are insoluble in it. Sand and dirt are two examples of these solids, and in the situation of vegetable oil seed fragments and Animal fats or used cooking oil can contain bone and gum fragments.

## ***Unsaponifiable Matter***

Organics are the primary constituents of unsaponifiable matter, in which the base does not react in order to produce soap. A Higher molecular weight of alcohols, waxes, sterols, pigments, and hydrocarbons are all included. After the transesterification reaction, these non-polar components are still present in the biodiesel.

## ***Moisture, Insoluble, and Unsaponifiable (MIU)***

It's the fat or oil that MIU represents, those that it is not possible to change to mono alkyl fatty esters by transesterification or esterification.

## ***Oxidation Stability***

Based on oxidation stability, age or the conditions in which it was stored of the oil or fat that can be determined and is able to foretell if the feedstock is accomplishing a goal of the lowest necessities for the stability of biodiesel oxidation, as stated by. Two factors influence the stability of oxidation, first the fact that hydrogen atoms exist, next to double bonds of carbon-carbon, in which oxidation can take place. Second is the presence of naturally occurring antioxidants in the raw material, can avoid of the triglyceride molecules oxidation.

## ***Magnesium and Phosphorous, Calcium***

The needs of phosphorous in biodiesel be partial to (10 ppm) (0.001 % mass maximum), and It is expected that the total amount of magnesium and calcium will be less than (5 ppm), minor components such as calcium, phosphorus, and magnesium are typically associated with gums and phospholipids that have the potential to function as emulsifiers or produce sediment, transesterification yields can be reduced.

## ***Transesterification***

Vegetable oils can be transformed into diesel-like fuels using a variety of methods. Transesterification, blending, cracking, micro emulsification, and pyrolysis are all examples of these processes. Biodiesel is most commonly produced through the transesterification process. When compared to other techniques, this method is more efficient. it has a wide range of benefits. e.g., Biodiesel of high quality is produced with a high yield in normal conditions. Over the last few years, transesterification has gained widespread acceptance for the converting vegetable oils into fuel-like products that are more compatible from a technical standpoint. To make biodiesel, transesterification is a crucial step in the production process so that it can bring the feedstock-vegetable oils closer to the viscosity of Diesel fuel that is derived from fossil fuels. Transesterification is a steady-state process,

describing the alcoholics of carboxylic esters usually performed in the in the presence of a standard catalyst, e.g., NaOH and KOH, for valued accelerate the rate at which the equilibrium adjustment is made to increase the amount of esters produced. An appropriate catalyst is needed for transesterification, in which alcohol is mixed with vegetable oil and reacted. Because ethyl and methyl alcohols are so widely used, ethyl-methyl esters are created. As a result of the response, liquids in two layers of distinct, i.e., ethyl-methyl ester and glycerin are appeared, which are separated. There are two types of transesterification: catalytic and noncatalytic processes.

There are many variables that affect the rate of biodiesel production via transesterification, of these, the most critical are type and amount of alcohol, type and amount of catalyst, reaction time and temperature. The yield is also affected by the feedstock's content of moisture and the free fatty acids amount. Generally, Reaction time is 2 hours at a temperature between 50 and 75 degrees Celsius. Gravity or centrifugal force is used to separate biodiesel from glycerol.

While transesterification has many benefits, it also has a few drawbacks to consider. Vegetable oil transesterification takes a long time to complete. and Only a few post-treatment steps, such as separation, are required, the use of water for washing and heating purposes. To purify biodiesel, more water is needed.

## **Catalyst**

Catalysts for biodiesel production generally fall into one of three categories: alkalis, acids, or enzymes. the used in biodiesel production Alkali and acid catalysts are more frequently than enzyme catalysts. Homogeneous and heterogeneous catalysts were the next classifications made. Enzyme catalysts, on the other hand, have recently gained popularity due to their ability to reduce soap formation and their ease of purification. Despite this, they aren't widely used because of their slower response times and higher costs. Despite this, they aren't widely used because of their slower response times (longer reaction times), and costs higher. Some researchers recently, new biocatalysts have been created, effort to reduce costs. These biocatalysts don't require purification, which is a benefit.

## **Homogeneous base-catalyst**

Using homogeneous alkaline catalysts for transesterification is more common due to the slower rate of the acid counterpart. Potassium hydroxide is the most commonly used basic catalyst (KOH), (KOCH<sub>3</sub>) potassium methoxide, (NaOH) sodium hydroxide, (NaOCH<sub>3</sub>) sodium methoxide, and sodium ethoxide (NaOCH<sub>2</sub>CH<sub>3</sub>). As a result of their abilities, to catalyze reactions at atmospheric pressure and low temperature, these catalysts are widely used, fast and efficient conversion rates, as well as a low cost.

(KOCH<sub>3</sub>) potassium methoxide and (NaOCH<sub>3</sub>) Sodium methoxide, are improved catalyst than (NaOH) sodium hydroxide and (KOH)potassium hydroxide, as a result of the dissociation into CH<sub>3</sub>O<sup>-</sup> and Na<sup>+</sup> and CH<sub>3</sub>O and K<sup>+</sup> respectively when comparing on the yield of biodiesel.<sup>46</sup>Commercial biodieselproduction typically employs an alkaline catalyst The reason for this is that it does not generate water duringthe transesterification reaction. The one that is most frequently used homogeneous base catalysts in the production of biodiesel are NaOH and KOH.

## **Homogeneous acid-catalyst**

in waste oils the Free fatty acids can't be converted to biodiesel using an alkaline catalyst. Ester separation will be prevented by soap derived from these FFAs, washing water, and glycerin, Once, with the help of an alkaline catalyst, the (liquid acid-catalyzed transesterification), is offered as a means of overcoming a lot of the conundrums related with liquid base catalysts. Sulfonic acid, hydrochloric acid, Sulfuric acid, organic sulfonic acid, for transesterification, acids containing ferric sulphate are the most widely used catalysts in chemical reactions.

## **Heterogeneous base-catalyst**

Heterogeneous catalysts have been the subject of numerous investigations in production of biodieselin order to get around the drawbacks of homogeneous catalysis. the vast majority of biodiesel production heterogeneous catalysts are whichever (alkaline oxide) or (alkaline earth metal oxide), that large area was successfully supported.



The activity of heterogeneous basic catalysts is greater than that of heterogeneous acid catalysts. Their homogenous counterparts are quite similar. Solid alkaline catalysts, for instance, (CaO) calcium oxide provide numerous benefits such as long catalyst life times, higher activity, and running in a moderate reaction condition is possible. Nonetheless, CaO as catalyst can also be used to reduce the rate of reaction in the biodiesel synthesis process.

### **Heterogeneous acid-catalyst**

However, despite the homogeneous acid catalyst's efficiency, can lead to problems contamination that necessitate effective separation and procedures for removing impurities from a product. As a result, production costs will rise. Heterogeneous acid catalysts are a viable alternative to homogeneous acid catalysts and have the potential to be more efficient. The heterogeneous acid catalysts have some advantages, insensitive to content of (FFA), can simultaneously conduct the esterification and transesterification, remove the (washing step) from the biodiesel production process, a more efficient way to separate of catalyst from product, reusing and regenerating the catalyst, is possible and also reduce the issues with corrosion. Solid acid catalysts are preferable over liquid acid catalysts in many applications, They contain a number of different sites as a result of this, with different strength of Bronsted, or Lewis acidity. A significant advantage of Bronsted acid catalysts based is that it promotes the simultaneous esterification and transesterification of fatty acids with the utilization of fewer expensive feedstock with an increased concentration of free fatty acids.

### **Enzyme (Biocatalyst) catalyst**

Researchers have focused on enzymatic transesterification, as a result of an issue with downstream processing posed through the process of transesterification in the lab. Massive amounts of wastewater are being produced, and glycerol recovery is problematic for a variety of reasons, that it will eventually rise the total cost of producing biodiesel and not being ecofriendly, In contrast, enzyme catalysis happens without the generation of (by-products), insensitive to high (FFA oil), easy recovery product and mild condition of reaction, and it is possible to re-use the catalyst. Enzyme catalyzed biodiesel production has a process with a high probability of being environmentally friendly and a talented alternative to the chemical process. Biodiesel production was aided by the enzyme has some limits particularly when implemented at a large-scale manufacturing facility as a result of the high cost of enzyme, and enzyme deactivation and slow reaction rate.

### **Biodiesel Properties**

Cetane number and cold flow characteristics are define biodiesel's properties, viscosity, oxidation stability, and fuel density. A fuel's ability to be pumped and filtered is its cloud point has influenced it and cold filter plugging point. regardless of Biodiesel sources, has a cloud point that is generally higher in comparison with diesel. Saturated fatty acid content of biodiesel may rise the cloud point. saturated fatty acids with long chains lead to biodiesel with higher cloud points. the biodiesel derived from less saturated fatty acid-containing sources has lower cloud points. Additives, on the other hand, the cold flow enhancers or transesterification procedures with branches of alcohol, can be put to use to biodiesel's cold flow properties should be improved.

is a measure of the fuel's cetane number is an important factor in determining its cold flow properties. The High cetane number fuels have ignition and combustion are separated by a short period of time. In spite of the fact that a low cetane number would not be ideal, this may also cause fuel to burn near the injection line if it has a high cetane number, which results in a clogged nozzle. Generally, the cetane number of conventional diesels is lesser than that of biodiesel, Cetane number can rise in the existence of Fatty acids with a long chain and a saturated fatty acids high content. For example, biodiesel from (UCO), has a higher cetane rating comparative to (SO) biodiesel. a link exists between biodiesel's cetane number and its viscosity. Therefore, the use of the (grease and AF) with high viscosity can result in biodiesel with a high cetane number. Combustion of an engine with a high cetane number may result in an increase in gaseous pollution. PM and CO emissions are still a concern, may decrease if preferable to use biodiesel with a high cetane number is used in the combustion process. NOx emissions

are rising, because of Biodiesel with higher oxygen content. The viscosity of biodiesel is another important factor in determining its quality. The injection and atomization processes will be impacted by high viscosity. Because of the large droplets that form, and poor mixing with air during the injection process, it may result in decreased combustion efficiency. The high viscosity of biodiesel is attributed to its saturated fatty acids at high content and long chains of carbon. Fuel injection and combustion in an engine are both impacted by density, and viscosity. Slow injection can be caused by a high density. Slow injection can be caused by a high density, as a result, the atomization and combustion process are less efficient. Further, Pollutant emissions could be accelerated as a result, from the engine. Generally speaking, the density of biodiesel is higher than that of conventional diesel fuel. Unsaturation at a high level, the number of double bonds, in carbon chains, and the density of biodiesel is amplified by the attendance of hydroxyl esters in the fatty acids.

Biodiesel's oxidation stability is yet another essential quality. The storage of biodiesel for an extended period of time can result in changes in the biodiesel quality due to the fatty acid's oxidation. Oxidation results in the formation of gums, which has a significant impact on the fuel's physiochemical properties (e.g., viscosity, cetane number, etc.) and This will lead to a less efficient combustion. During the process of burning oxidized fuel, carbon deposits form in the combustion chamber. Nonoxidized biodiesel produces less NO<sub>x</sub> than oxidized biodiesel. A biodiesel's oxidation tendency rises as the unsaturation level of fatty acids rises. Consequently, biodiesel derived with linoleic and linolenic acid concentrations in vegetable oils above normal, such as SO biodiesel, has a low oxidation stability. " The temperature point at which the fuel supply begins to ignite is known as the "flash point," and it is closely linked to the volatility of the fuel. the fuel with a high flash point indicates delay combustion as a result of in which form does carbon deposition take place, and large-scale emissions of polluting gases (e.g., NO<sub>x</sub>), Long carbon chain fatty acids are present in biodiesel, make it less volatile and a higher flash point, compared to diesel fuel in its traditional form.

### **Performance of biodiesel in diesel engine**

Biodiesel it has shown promise as an alternative diesel engine fuel in numerous studies. However, For the most part, vegetable oils or oil mixes when utilised directly are considered undesirable and unworkable, for both direct and indirect injection diesel engines. Viscosity, acidity, and the presence of free fatty acids of such oils, along with the formation of gums, because of the oxidation and polymerization that occurs during storage and A few of the more obvious issues include combustion, oil thickening, carbon deposits.

Combustion of Biodiesel in a Diesel Engine, several promising characteristics have been demonstrated, reduction of exhaust emissions is also included. Because of their similar properties to diesel, trans esterified oils are a promising alternative to the fossil fuel. Petroleum-based diesel fuel has the same chemical and physical properties as biodiesel, Because of this, it is suitable for use in diesel engines. Existing diesel engines can use biodiesel as a diesel fuel replacement with no modifications needed. In accordance with the results, Compression ignition engines can use biodiesel without making any adjustments, in place of diesel. Fueled by the same principles as petroleum diesel, Compression-ignition engines are required for biodiesel use. Blends of up to 20 biodiesels mixed with petroleum diesel fuels, can be used in a wide range of diesel engines, and compatible with a wide range of storage systems and distribution systems.

### **Reviews from Previous studies:**

Md Mofijur Rahman et al., (2017) investigated the wear and frictional properties of native Australian species, biodiesels of the first and second generation using a four-ball tribotester. It was decided to conduct the tribological test at 1800 rpm with various temperatures and loads. The shape of the ball's exterior and pattern on the surface was tested by (SEM) scanning electron microscope and (EDX) energy dispersive X-ray spectroscopy analysis. The results of the experiments demonstrated that the usage of biodiesel has (COF) a lower friction coefficient, and lower (WSD) wear scar diameter up to (83.50% and 41.28%), respectively when compared to regular diesel. The results of the worn-surface area measurements revealed that A biodiesel fuel must contain a minimum amount of carbon and oxygen, except Fe, compared to diesel. Also discovered was a worn surface area for diesel that

(2.20%–27.92%) to be higher than biodiesel. The results of this investigation showed that the tribological performance of examines the total cost of production of the biofuel blends; was better more so than diesel, and among the various types of biodiesel fuel, macadamia biodiesel displayed enhanced lubrication properties.

Azad et al., (2013) investigated experimentally the use of a diesel direct-injection engine, with 1st generation biofuel blends with fossil fuel, like kerosene. Trans-esterification reactions have not been used to produce the first generation biofuel (mustard oil). M20, M30, M40, and M50 are the approximate volume percentages used in the blend with kerosene oil. The biofuel's characteristics blends were tested in accordance with a variety of ASTM standards. It was determined and compared how temperature changes the properties of various biofuels. After that, a four-stroke test engine, single cylinder, in place of the bed of the dynamometer used to measure a variety of parameters of the engine, like break house power, and break specific fuel consumption, break mean effective pressure, and lube oil temperature, temperature and volume of exhaust gas etc. A thorough investigation into the performance of various biodiesel blends was conducted with kerosene and performance of diesel fuel in engines. This paper also discusses the biofuel blends' total cost of production per unit and their economic impact. Emilio A. Viorneri Portillo et al., (2020) carried out a comparative assessment of Environmental impacts of (B25) biodiesel blend production and use, and Using ultra-low Sulphur fuel in a power generator of (33 kW) at 100 percent of the rated impact and load analysis based on the lifecycle methodology, Based on primary data for each stage of processing and utilization. The model's final output, that (B25) overtakes low Sulphur diesel in every class. Abiotic depletion was found to have the greatest impact reductions, potentially harmful to human health, acidification potential, Classifications of eutrophication and global warming with (39.48%), (39.44%), (39.24%), (38.73%), and (35.77%) respectively, Exhaust emission measurements carried out in the laboratory a rise was observed in (CO) (52%) and a decrease in (NO<sub>x</sub>) (41.54%) for (B25), As compared to diesel with ultralow Sulphur content. The findings demonstrate that biodiesel is beneficial and a viable alternative fuel.

Hirani et al., (2018) focused on the most recent developments in scientific study of the second-generation biofuel production from diverse feedstocks. reducing emissions of greenhouse gases and the dependence on fossil fuels are two of the primary goals of renewable sources of energy. Biofuel industries, particularly ethanol and biodiesel, have been rapidly expanding alongside agricultural crop production for more than a decade. Agriculture crops are heavily relied upon in the production of the first-generation biofuels such as corn, sugarcane, sugar beets, soybean, and canola. As a result, for the past few years, the inherent competition between food and fuel has been up for debate in our society. Next-generation biofuels, many different feedstocks can be used to produce, including agricultural waste products, recent technological advances in research and development, high-yielding grass species' cellulosic biomass and crop residues.

Cristiano Varrone et al., (2017) investigation of various mixed microbial cultures' performance (MMC), capable of fermenting glycerol crude produced using biodiesel produced from the fat of animals, to produce (1,3) propanediol (1,3PDO), and During non-sterile conditions (butyric acid). There were eight different CSTRs, each with a different inoculum and growth medium. Each CSTR was set up in a different way. The metabolic product distribution was studied under varying operating conditions. The kinetic properties of each MMC were analyzed, and stoichiometric reactions were designed for each of the reactions in the dataset. The use of Next Generation Sequencing (NGS) allowed researchers to examine changes in microbial populations. The maximum rate of degradation of the substrate was about (101 g/L/d) of glycerol, with an output of (38 g/L/d and 11 g/L/d) for 1,3 PDO and butyric acid respectively, the results were obtained using a hydraulic retention duration of (12 h) and 60 g/L input. A concentration that is very close to the maximum of feedstock was achieved, (90 g/L), resulting in an incomplete degradation of the substrate.

Dariusz Kurczyński et al., (2021) studied the exhaust gas composition as a result of second-generation biofuels, from a Perkins diesel engine. Diesel fuel and first-generation biofuel were used as a standard for comparison. Biofuel of the second generation, the babassu butyl esters (BBuE) are created through transesterification with the use of a sulphuric acid catalyst, and butyl alcohol. Rapeseed oil methyl esters were used as the biofuel of first



generation in the tests (RME). A component of the engine's combustion process as a concentrations of carbon monoxide, carbon dioxide, hydrocarbons and amount of particulate matter in the exhaust was lowered in the engine, as a comparison that powered by either (RME or DF). By using BBUe as the engine's power source, lower nitrogen oxide concentrations were found in comparison to its powering with the (RME). Since (BBUe) contains several short chain esters that are not present in (RME). This has led to the development of a brand-new type of biofuel powered engine has fewer potentially harmful components emitted into the atmosphere as exhaust gas than do engines powered by first generation biofuel or diesel oil.

## Conclusion

However, the high cost of these oils and fats adds up to a higher total cost of production. The improvement of biodiesel quality with less harm to the environment has been the primary focus of successive generations' research. Biodiesel feedstocks for the fourth generation are still in the early stages of development. Uses synthetic biology technology to produce biodiesel in the fourth generation, which will be the next generation of biodiesel. This generation uses metabolic engineering, biological tools should be added to the enhancement of biodiesel's quality and quantity from various feedstocks. The content energy of biodiesel, from the fourth generation is especially high, rapid of feedstock growth, and amplified capacity for CO<sub>2</sub> absorption during the process of creating feedstock, as an alternative, several research projects have been carried out, in order to advance catalysts at a reasonable price, that reduce the cost of production. a variety of catalysts, such as homogeneous or heterogeneous acid catalysts, the Researchers have explored, and used biocatalysts (enzymes), and heterogeneous-homogeneous base catalysts for biodiesel synthesis. In commercial production, base-catalyzed transesterification is frequently employed, because of the high yield of FAME, in low activation energy requirements with a short reaction time as compared to acid-catalyzed transesterification. a method of transesterification that uses an acid catalyst has been reported requirements higher molar ratios of alcohol to oil, it is the most effective way when a poor quality product of oil is used From the investigation, This means that It is possible to utilise biodiesel as a substitute for diesel fuel in a compression ignition engine, without modifying the engine in any way. In the same way that petroleum diesel works, Compression-ignition engines can run on biodiesel

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